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By Christiane Gottschalk;Judy Ann Libra;Adrian Saupe

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<http://www.tandfonline.com/doi/pdf/10.1080/00207233.2010.517628>

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1. Water Res. 2003 Apr;37(7):1443-67. Ozonation of drinking water: part I. Oxidation kinetics and product formation. von Gunten U. Swiss Federal Institute for
<http://www.ncbi.nlm.nih.gov/pubmed/12600374>

Water Treatability Database | US EPA -

Ozone (O₃) is one of the strongest disinfectants and oxidants available in drinking water treatment. Ozone must be generated onsite and used immediately.
<http://iaspub.epa.gov/tdb/pages/treatment/treatmentOverview.do?treatmentProcessId=-1467636837>

Ozonation of drinking water: Part II -

This is the second part of an article on drinking water ozonation processes dealing with disinfection and formation of disinfection by-products.
<http://www.sciencedirect.com/science/article/pii/S004313540200458X>

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(drinking water), but water purification Another advantage of ozone is that it leaves no residual disinfectant in the water. Ozone has been used in drinking
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<http://onlinelibrary.wiley.com/doi/10.1002/9783527628926.ch3/summary>

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Water: A Matrix of Life , Christiane Gottschalk, Judy Ann Libra, Adrian Saupe) , Handbook of Ozone Technology and Applications ,

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